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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$ R factor = 0.036 wR factor = 0.109 Data-to-parameter ratio = 12.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-Oxo-2,3-dihydro-1*H*-indol-3-one nicotinoylhydrazone

In the title structure, $C_{14}H_{10}N_4O_2$, the pyridyl ring is twisted with respect to the central -C(=O)-NH-N= fragment by 17.7 (1)°, whereas the indole fused-ring system is almost coplanar with this fragment. Molecules are linked into a linear chain motif by $N-H \cdots O$ hydrogen bonds.

Comment

The crystal structure of isatin 2-picolylhydrazone, (II), has been determined as part of a report on the biological activity of its transition metal derivatives. The molecule adopts a nearly planar conformation, the planar nature being conducive to N,N',O-chelation in the deprotonated compound (Rodríguez-Argüelles *et al.*, 2004).

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The isomeric title molecule, (I) (Fig. 1), adopts a somewhat non-planar conformation. The pyridyl ring is twisted by 17.7 (1)° with respect to the central -C(=O)-NH-N=fragment, while the indole fused-ring system is nearly coplanar with this fragment [4.1 (1)°]. The bond distances and angles in (I) and (II) are the same within experimental error. In the crystal structure of (I), molecules are linked into a linear chain by N-H···O hydrogen bonds (Table 1 and Fig. 2). Transition metal derivatives of the title compound have been reported, but these have not been crystallographically verified (Srivastava *et al.*, 2000).



Figure 1

ORTEPII plot (Johnson, 1976) of the title molecule. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii.

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Experimental

Isatin (0.50 g, 3.39 mmol) and nicotinoyl hydrazide (4.66 g, 3.39 mmol) were dissolved in ethanol (30 ml); the solution was heated under reflux for 2 h. The orange solid that was isolated when the solution was cooled was collected and recrystallized from dimethyl-formamide to afford block crystals.

 $D_x = 1.452 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 5073

reflections

 $\mu = 0.10 \text{ mm}^{-1}$

T = 295 (2) K Block, orange $0.37 \times 0.35 \times 0.32$ mm

 $R_{\rm int} = 0.020$

 $\begin{array}{l} \theta_{\rm max} = 27.0^{\circ} \\ h = -11 \rightarrow 11 \end{array}$

 $k = -14 \rightarrow 14$

 $l = -15 \rightarrow 14$

 $\theta = 2.3 - 27.0^{\circ}$

Crystal data

$C_{14}H_{10}N_4O_2$
$M_r = 266.26$
Monoclinic, $P2_1/c$
a = 8.9066 (5) Å
b = 11.7021 (6) Å
c = 11.7768 (6) Å
$\beta = 97.097 \ (1)^{\circ}$
$V = 1218.0 (1) \text{ Å}^3$
Z = 4

Data collection

Bruker SMART area-detector diffractometer φ and ω scans 10102 measured reflections 2643 independent reflections 2149 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0617P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.036$	+ 0.2785P]
$wR(F^2) = 0.109$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} = 0.001$
2643 reflections	$\Delta \rho_{\rm max} = 0.23 \ {\rm e} \ {\rm \AA}^{-3}$
221 parameters	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$
All H-atom parameters refined	

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	<i>D</i> -H	Н∙∙∙А	$D \cdots A$	$D - H \cdots A$
$\frac{N2-H2n\cdots O2}{N4-H4n\cdots O1^{i}}$	0.86(1)	2.00 (1)	2.711 (2)	139 (2)
	0.86(1)	2.15 (1)	2.941 (2)	153 (2)

Symmetry code: (i) $x, \frac{1}{2} - y, \frac{1}{2} + z$.



Figure 2

ORTEPII plot (Johnson, 1976) of the hydrogen-bonded chain. Hydrogen bonds are indicated by dashed lines.

The positional parameters of all H atoms were refined with restraints of N-H = 0.86 (1) Å and C-H = 0.95 (1) Å; the isotropic displacement parameters were refined without restraint.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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References

Bruker (2001). SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.

Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Rodríguez-Argüelles, M. C., Ferrari, M. B., Bisceglie, F., Pelizzi, C., Pelosi, G., Pinell, S. & Sassi, M. (2004). J. Inorg. Biochem. 98, 313–321.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Srivastava, A. K., Pandey, O. P. & Sengupta, S. K. (2000). Synth. React. Inorg. Met. Org. Chem. 30, 1405–1416.